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Modern Chemistry Chapter 15 Test Review 25 multiple choice definition of self-ionization  $[H_3O^+][OH^-] = 1 \times 10^{-14}$  definition and formula for calculating  $pH = -\log[H_3O^+]$   $pH > 7$  is a base;  $pH < 7$  is an acid;  $pH = 7$  is neutral  $pH$  range is normally 0 to 14 calculate  $pH$  from  $[H_3O^+]$  calculate  $[H_3O^+]$  from  $pH$  definitions of indicators, transition interval, titration, neutralization, endpoint, standard solution acid-base titration monitors  $pH$  titration problems using  $V_a M_a e_{aq} = V_b M_b e_{aq}$ .

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## Modern Chemistry Chapter 15 Acid-Base Titration & pH

Modern Chemistry 136 Chapter Test Name Class Date Chapter Test A, continued  
\_\_\_\_14. A  $pH$  meter measures the  $pH$  of a solution by measuring the a. amount of

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charge on the hydronium ion. b. color of the solution. c. concentration of the solution. d. voltage difference between two electrodes in the meter's probe.

\_\_\_\_\_15. When you conduct an

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Chemistry Chapter 15 Test - seapa.org Modern Chemistry Chapter 15 Test Review  
25 multiple choice definition of self-ionization  $[H_3O^+] [OH^-] = 1 \times 10^{-14}$  definition  
and formula for calculating  $pH = -\log[H_3O^+]$   $pH > 7$  is a base;  $pH < 7$  is an acid;  $pH$   
 $= 7$  is neutral  $pH$

### **Modern Chemistry Chapter 15 Test**

Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued  
18. When a strong acid is titrated with a weak base, the  $pH$  of the solution at the  
equivalence point is than 7. 19. A is a highly purified solid used to check the  
concentration of a standard solution. 20. A 1 M solution of NaOH will have a  $pH$   
that is

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modern chemistry chapter 15. STUDY. PLAY. self ionization of water. two water molecules produce a hydronium ion and a hydroxide ion through proton transfer.  $\text{H}_2\text{O}(\text{l}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{H}_3\text{O}^+(\text{aq}) + \text{OH}^-(\text{aq})$  concentration of  $\text{H}_3\text{O}^+$  and  $\text{OH}^-$  in water at 25 degrees celcius.  $1.0 \times 10^{-7}$  mol/L. ionization constant of water.

## **Modern Chemistry: Chapter 14 & 15 Flashcards | Quizlet**

Modern Chemistry: Chapter 14 & 15. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. DorettaA TEACHER. Terms in this set (49) Binary Acid. an acid that does not contain oxygen, such as hydrofluoric acid. Oxyacid. an acid that is a compound of hydrogen, oxygen, and a third element usually a nonmetal.

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